Stereoselective Synthesis of cis-2,5-Disubstituted Tetrahydrofurans Using Oxabicyclo[3.2.1]heptanone **Platforms. Building Blocks for Natural Product Synthesis**

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Versatile syntheses of cis-2,5-disubstituted tetrahydrofurans represent an important challenge because of the presence of this structural unit in polyoxygenated terpenes such as eurylene (1),¹ polyether antibiotic ionophores such as ionomycin (2),² and other natural products (Figure 1). Methods which allow the stereoselective introduction of functionality contiguous to C-2 and C-5 (i.e., at C-1' and C-1") are particularly attractive because such fragments (e.g., 3) are potentially useful building blocks in natural product synthesis. Among the numerous approaches to this class of compound, perhaps the most successful are those which utilize 1,5-dienes or substituted pentenols as starting materials.³ These methods are limited by the availability of suitable alkene precursors. Consequently, development of new synthetic methods for the preparation of tetrahydrofurans continues to attract attention.⁴

Methods in which oxabicyclic systems are used as a stereochemical scaffold to prepare functionalized tetrahydrofurans provide an attractive alternative route to these important ring systems.⁵ We recently described highly diastereoselective, Lewis acid catalyzed, [3 + 4] and [3+ 5] annulation procedures in which 1,4- and 1,5dicarbonyl compounds reacted with bis(trimethylsilyl) enol ether 5 to provide oxabicyclo[3.2.1]octanones and oxabicyclo[3.3.1]nonanones, respectively.⁶ We envisaged that the stereoselective manipulation of 6 (Scheme 1), prepared using the [3 + 4] annulation of 4 and 5, could lead to a highly versatile tetrahydrofuran synthesis. Bicyclic ether 6 contains two of the four stereocenters



Eurylene (1)



lonomycin (2)





present in fragment 3, and we reasoned that the other two stereocenters, corresponding to C-2 and C-4 in 6, could be introduced through the selective manipulation of the β -keto ester functionality present in **6**. Our results describing the successful accomplishment of these goals are described below.

Results and Discussion

Bicyclic ether 6 was prepared (55%) according to the published procedure (Scheme 2).⁶ Stereoselective alkylation of the derived β -keto ester enolate afforded keto ester 7 (92%) as the only isolated product. Baeyer-Villiger oxidation of 7 initially proved troublesome. Treatment with m-CPBA at room temperature for 24 h led to only 8% conversion of the starting material. The use of elevated temperature and a radical scavenger gave the desired lactone 8 in 49% yield after 86 h.7 Trifluoroperacetic acid prepared using urea hydrogen peroxide as an anhydrous source of peroxide also failed to give the product in an acceptable yield.⁸ The rate enhancement of Baeyer-Villiger oxidations by NaHCO3 has been reported previously,⁹ and treatment of 7 with *m*-CPBA and NaHCO₃ led to a faster, regioselective reaction in which lactone 8 was obtained in 67% yield (89% based on recovered starting material) after 72 h. Subsequent hydroxylation using Davis's protocol¹⁰ afforded the α -hydroxy lactone 10 as a single diastereoisomer (48%) along with a number of byproducts including the bislactone 9

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^{(2) (}a) Polyether Antibiotics: Naturally Occuring Acid Ionophores; Westley, J. W., Ed.; Marcel Decker: New York, 1983; Vol. 1, 2. (b)

<sup>Robinson, J. A. Prog. Chem. Org. Nat. Prod. 1991, 58, 1.
(3) For a review of approaches to cis-2,5-disubstituted tetrahydro-</sup>

⁽³⁾ For a review of approaches to cis-2,5-disubstituted tetrahydro-furans, see: Boivin, T. L. B. Tetrahedron 1987, 43, 3309.
(4) For some recent examples, see: (a) Clark, J. S. Tetrahedron Lett. 1992, 33, 6193. (b) Feldman, K. S.; Fischer, T. E. Tetrahedron 1989, 45, 2969. (c) Craig, D.; Smith, A. M. Tetrahedron Lett. 1992, 33, 695.
(d) Lipshutz, B. H.; Barton, J. C. J. Am. Chem. Soc. 1992, 114, 1084.
(e) Barrett, A. G. M.; Flygare, J. A.; Spilling, C. D. J. Org. Chem. 1989, 54, 4723. (f) Spino, C.; Weiler, L. Tetrahedron Lett. 1987, 28, 731. (g) Walka, D. M. Drurbula, C. A. & Walkar, C. B. La Mar, Chem. Soc. Walba, D. M.; Przybyla, C. A.; Walker, C. B., Jr. J. Am. Chem. Soc. 1990, 112, 5624. (h) Paterson, I. A.; Craw, P. A. Tetrahedron Lett. **1989**, 30, 5799. (i) Schreiber, S. L.; Sammakia, T.; Hulin, B.; Shulte, G. J. Am. Chem. Soc. **1986**, 108, 2106. (j) Still, W. C.; Romero, A. G. J. Am. Chem. Soc. **1986**, 108, 2105. (5) (a) Mann, J. Tetrahedron **1986**, 42, 4611 and references therein.

⁽b) Grese, T. A.; Hutchinson, K. D.; Overman, L. E. J. Org. Chem. **1993**, 58, 2468. (c) Evans, D. A.; Dow, R. L.; Shih, T. L.; Takacs, J. M.; Zahler, B. J. Am. Chem. Soc. 1990, 112, 5290. (d) Tochtermann, W.; Olsson,
 G.; Peters, E.-M.; Peters, K.; Von Schnering, H. G. Tetrahedron 1988,
 44, 4797. (e) Noyori, R.; Sato, T.; Hayakawa, Y. J. Am. Chem. Soc.
 1978, 100, 2561. (f) Sato, T.; Watanabe, M.; Noyori, R. Chem. Lett. 1980, 679

^{(6) (}a) Molander, G. A.; Cameron, K. O. J. Am. Chem. Soc. 1993, 115, 830. (b) Molander, G. A.; Cameron, K. O. J. Org. Chem. 1993, 58. 5931.

⁽⁷⁾ Grieco, P. A.; Oguri, T.; Gilman, S. J. Am. Chem. Soc. 1980, 102, 5886

⁽⁸⁾ Cooper, M. S.; Heaney, H.; Newbold, A. J.; Sanderson, W. R. Synlett 1990, 535. (9) Whitesell, J. K.; Mathews, R. S.; Helbling, A. M. J. Org. Chem.

^{1978, 43, 784.} (10) Davis, F. A.; Vishwakarma, L. C.; Billmers, J. M.; Finn, J. J. Org. Chem. 1984, 49, 3243.



^a Key: (a) cat. TMSOTf, 5, CH₂Cl₂, -78 °C; (b) NaH, MeI, THF, -10 °C; (c) m-CPBA, NaHCO₃, (CH₂Cl)₂; (d) KHMDS, transbenzenesulfonyl)-3-phenyloxaziridine, THF, -78 °C; (e) TBDM-SOTf, 2,6-lutidine, CH₂Cl₂, 0 °C to rt; (f) DIBALH, THF, -78 °C to rt; (g) DBU, CH₂Cl₂.

(10%). The α -orientation of the hydroxyl group in 10 was readily determined by X-ray analysis.¹⁷ An examination of models¹¹ suggested that the si face of the enolate, derived from 8, would be blocked by the presence of the β -methyl group at C-2 and that enolate oxidation would occur predominantly from the re face. Stereoelectronic effects may also contribute to the high diastereoselectivity observed.¹² Attempts to oxidize the enolate derived from 8 using MoO_5 ·Py·DMPU¹³ as the oxidizing agent gave only recovered starting material. Protection of the alcohol 10 as its TBDMS ether 11 (99%) followed by treatment with DIBALH afforded monoprotected tetrol 12 (73%). DBU-catalyzed migration of the TBDMS protecting group from the secondary to the adjacent primary hydroxyl gave the key intermediate 13 (82%, 98% based on recovered starting material).14

Both 12 and 13 possess the four stereocenters in the core tetrahydrofuran unit required in the synthesis of several natural product fragments, and all four hydroxyl groups in 12 can be readily differentiated using standard synthetic methods and converted to versatile intermediates for further elaboration. Perhaps most importantly, both ends of the tetrahydrofuran core can be selectively activated by conversion into epoxides 15 and 17, which are ideally suited for the sequential introduction of suitably functionalized side chains. Thus, protection of the 1,2-diol in 13 as the acetonide (96%) followed by mesylation of the remaining unprotected hydroxyl group afforded 14 (92%) (Scheme 3). Desilylation and in situ treatment with K_2CO_3 afforded epoxide 15 (98%). In this way C-2" is activated toward nucleophilic attack and the



 $^{\alpha}$ Key: (a) Me_2C(OMe)_2, cat. PPTS, CH_2Cl_2; (b) MsCl, cat. DMAP, pyridine; (c) Bu_4NF, THF, K_2CO_3; (d) MsCl, cat. DMAP, pyridine; (e) K₂CO₃, MeOH.

desired stereochemistry at C-1" is incorporated. Dimesylation of triol 13 afforded alcohol 16 (91%) which was readily converted to the epoxide 17 (97%) upon treatment with K_2CO_3 in MeOh, thereby activating C-2' for nucleophilic attack.

Not only are epoxides 15 and 17 synthetically useful but the potential to vary substituents at C-2 and C-5 in 10 by use of alternative electrophiles in the two enolate anion reactions make this a desirable approach to tetrahydrofuran synthesis. The ability to introduce substituents stereoselectively within the tetrahydrofuran ring using substituted 1,4-dicarbonyl compounds (c.f., 4) is also attractive.⁶

Experimental Section

Reagents. THF was distilled immediately prior to use from sodium benzophenone ketyl under argon. CH₂Cl₂ was distilled from CaH₂ and stored over 4-Å molecular sieves. m-CPBA was purified according to the literature procedure.¹⁵ Standard benchtop techniques were employed for handling air-sensitive reagents.16

 $(1R^*, 2S^*, 5S^*)$ -2,5-Dimethyl-2-(methoxycarbonyl)-8oxabicyclo[3.2.1]octan-3-one (7). Sodium hydride (58 mg of a 60% suspension in mineral oil, 1.45 mmol) was washed under argon with dry hexanes, suspended in THF (2 mL) at rt, and treated dropwise with a solution of 6 (260 mg, 1.31 mmol) in THF (4 mL). After 20 min the solution was cooled to -10 °C and methyl iodide (340 mg, 2.40 mmol) was added dropwise. After 3 h the solution was concentrated in vacuo and the residue was dissolved in ether and filtered through Celite. Concentration in vacuo followed by flash chromatography (7:3 hexanes/ ethyl acetate) and Kugelrohr distillation provided 7 as a white solid (251 mg, 90%): mp 43-45 °C; bp 80-85 °C/0.3 mmHg; ¹H NMR (300 MHz, CDCl₃) δ 4.38 (dd, J = 6, 3 Hz, 1H), 3.71 (s, 3H), 2.63 (dd, J = 16, 2 Hz, 1H), 2.25 (d, J = 16 Hz, 1H), 2.15 (m, 2H), 1.85-1.60 (m, 2H), 1.56 (s, 3H), 1.42 (s, 3H); ^{13}C NMR $(75 \text{ MHz}, \text{CDCl}_3) \ \delta \ 205.6, \ 171.3, \ 82.3, \ 80.8, \ 61.9, \ 52.2, \ 51.2, \ 35.3,$ 28.4, 25.7, 21.1; IR (neat) 1730, 1716 cm⁻¹; HRMS calcd for C11H16O4 212.1049, found 212.106; LRMS (EI) m/z 212. Anal. Calcd for C₁₁H₁₆O₄: C, 62.25, H, 7.60. Found: C, 62.18; H, 7.54.

⁽¹¹⁾ Models were generated using Macromodel.

⁽¹²⁾ Deslongchamps, P. Stereoelectronic Effects in Organic Chem-(12) Desing in ange, 1. Subverse to the Effects in Organic Chemi-istry; Pergamon: Oxford, 1983; p 283.
(13) Anderson, J. C.; Smith, S. C. Synlett 1990, 107.
(14) (a) Torisawa, Y.; Shibasaki, M.; Ikegami, S. Tetrahedron Lett.

^{1979, 1865. (}b) Jones, S. S.; Reese, C. B. J. Chem. Soc., Perkin Trans. 1 1979, 2762.

⁽¹⁵⁾ Schwartz, N. N.; Blumbergs, J. H. J. Org. Chem. 1964, 29, 1976.
(16) Brown, H. C. Organic Syntheses via Boranes; Wiley: New York, 1975.

⁽¹⁷⁾ The authors have deposited atomic coordinates for structure 10 with the Cambridge Crystallographic Data Centre. The coordinates can be obtained, on request, from the Director, Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge, CB2 IEZ, UK.

(1R*,2S*,6S*)-2,6-Dimethyl-2-(methoxycarbonyl)-3,9dioxabicyclo[4.2.1]nonan-4-one (8). A stirred solution of 7 (5.0 g, 23.6 mmol) in $ClCH_2CH_2Cl$ (35 mL) was treated with NaHCO3 (2.9 g, 35.4 mmol) and m-CPBA (6.0 g, 35.4 mmol) and stirred at rt for 24 h. More NaHCO3 (2.9 g, 35.4 mmol) and m-CPBA (6.0 g, 35.4 mmol) were added, and the mixture was stirred for a further 48 h. The solution was diluted with CH2- Cl_2 and treated with a 10% solution of Na_2SO_3 followed by vigorous stirring for 15 min. The aqueous phase was extracted with CH₂Cl₂, and the combined organic phase was washed with saturated NaHCO3 and brine and dried over MgSO4. Concentration in vacuo followed by flash chromatography (3:1 hexanes/ ethyl acetate) afforded starting material (0.9 g, 18%) and 8 as a white solid (3.605 g, 67%): mp 96-97 °C; ¹H NMR (300 MHz, CDCl_3) δ 4.59 (d, J = 8.5 Hz, 1H), 3.76 (d, J = 1 Hz, 3H), 2.94 (d, J = 16 Hz, 1H), 2.82 (d, J = 16 Hz, 1H), 2.20-1.90 (m, 3H),1.80-1.70 (m, 1H), 1.73 (d, J = 1 Hz, 3H), 1.36 (s, 3H); ${}^{13}C$ NMR (75 MHz, CDCl₃) & 170.8, 170.2, 86.1, 81.6, 80.8, 53.1, 51.2, 34.7, 27.1, 26.3, 21.6; IR (CHCl₃) 1729 cm⁻¹; HRMS calcd for C11H16O5: 228.0998, found 228.0989; LRMS (EI) m/z 228. Anal. Calcd for C₁₁H₁₆O₅: C, 57.91; H, 7.07; Found: C, 57.75; H, 7.01.

(1R*,2S*,5S*,6S*)-1,5-Dimethyl-9-oxo-8,10,11-trioxatricyclo-[4.2.2.1^{2,5}]undecan-7-one (9) and (1R*,2S*,5S*,6S*)-2,6-Dimethyl-5-hydroxy-2-(methoxycarbonyl)-3,9-dioxabicyclo-[4.2.1]nonan-4-one (10). A solution of KHMDS (2.61 g, 13.1 mmol) in THF (325 mL) at -78 °C under argon was treated dropwise with a solution of 8 (2.0 g, 8.77 mmol) in THF (25 mL). After 20 min the solution was treated with trans-(benzenesulfonyl)-3-phenyloxaziridine (3.57 g, 13.8 mmol) in THF (260 mL) and stirred for a further 30 min. Camphorsulfonic acid (3.04 g, 13.1 mmol) in dry THF (30 mL) was added, and the solution was stirred for 30 min before being warmed to room temperature. Water was added, and the aqueous phase was extracted with ethyl acetate. The combined organic phase was washed with water, saturated NaHCO3, and brine and dried over MgSO4. Concentration in vacuo followed by flash chromatography (4:1 hexanes/ethyl acetate) afforded 9 as a white solid (190 mg, 10%) [mp 106-107 °C; ¹H NMR (300 MHz, CDCl₃) δ 4.68 (s, 1H), 4.12 (d, J = 7 Hz, 1H), 2.28 (m, 2H), 1.87 (m, 1H), 1.72 (m, 1H), 1.54(s, 3H), 1.53 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 168.5, 164.2, 85.1, 82.9, 82.4, 79.9, 29.9, 27.8, 22.9, 20.2; IR (neat) 1770 cm⁻¹; HRMS calcd for C10H12O5 212.0685, found 212.0683; LRMS (EI) m/z 212. Anal. Calcd for C₁₀H₁₂O₅: C, 56.60; H, 5.70. Found: C, 56.38; H, 5.73] followed, after recrystallization from ethyl acetate, by 10 as a white solid (1.02 g, 48%): mp 121-122 °C; ¹H NMR (300 MHz, CDCl₃) δ 4.59 (dd, J = 9, 2 Hz, 1H), 4.06 (t, J = 2 Hz, 1H), 3.77 (s, 3H), 3.68 (d, J = 2 Hz, 1H), 2.25-1.80 (m, 3H), 1.71 (s, 3H), 1.50–1.65 (m, 1H), 1.44 (s, 3H); ^{13}C NMR $(75 \text{ MHz}, \text{CDCl}_3) \delta 174.7, 170.1, 88.0, 83.9, 80.8, 76.6, 53.2, 31.5,$ 25.6, 24.0, 21.7; IR (neat) 3697, 1771, 1715 cm⁻¹; HRMS calcd for C11H16O6 244.0947, found 244.0948; LRMS (EI) m/z 244. Anal. Calcd for C₁₁H₁₆O₆: C, 54.09; H, 6.60. Found: C, 54.12; H, 6.68

(1R*,2S*,5S*,6S*)-5-[(tert-Butyldimethylsilyl)oxy]-2,6dimethyl-2-(methoxycarbonyl)-3,9-dioxabicyclo[4.2.1]nonan-4-one (11). A stirred solution of 10 (100 mg, 0.41 mmol) in CH₂Cl₂ (2 mL) at 0 °C under argon was treated with TBDMSOTf (102 mg, 0.46 mmol) followed by 2,6-lutidine (49 mg, 0.46 mmol). After 2 h the solution was diluted with CH2- Cl_2 and a saturated NaHCO_3 solution was added. The aqueous phase was extracted with CH2Cl2, and the combined organic phase was washed with cold 1% HCl, saturated NaHCO₃, and brine and dried over MgSO4. Concentration in vacuo followed by flash chromatography (9:1 hexanes/ethyl acetate) afforded 11 as a white solid (146 mg, 99%): mp 51-52.5 °C; ¹H NMR (300 MHz, CDCl₃) δ 4.57 (dd, J = 8.5, 2.5 Hz, 1H), 4.04 (d, J = 1.2 Hz, 1H), 3.76 (s, 3H), 2.20-1.80 (m, 3H), 1.69 (s, 3H), 1.65-1.50 (m, 1H), 1.39 (s, 3H), 0.89 (s, 9H), 0.13 (s, 3H), 0.01 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 172.1, 171.0, 86.1, 84.7, 81.1, 79.9, 53.0, 32.1, 26.0, 25.8, 25.2, 22.3, 18.4, -4.5, -5.8; IR (neat) 1738 cm^{-1} ; HRMS calcd for $C_{17}H_{30}SiO_6$ 358.1824, found 358.1812; LRMS (EI) m/z 358. Anal. Calcd for C17H30SiO6: C, 56.95; H, 8.44. Found: C, 56.88; H, 8.29.

 $(1''R^*, 1'R^*, 2S^*, 5R^*)$ -5-(1', 2'-Dihydroxy-1'-methylethyl)-5methyl-2-[1''-[(tert-butyldimethylsilyl)oxy]-2"-hydroxyethyl]tetrahydrofuran (12). A stirred solution of 11 (109 mg, 0.302 mmol) in THF (4 mL) at -78 °C under argon was treated with DIBALH (0.241 g, 1.508 mmol) and allowed to warm to rt. After 2 h a saturated solution of Rochelle's salts was added and the solution was diluted with ethyl acetate and stirred vigorously overnight. The aqueous phase was extracted with ethyl acetate, and the combined organic phase was dried over MgSO₄. Concentration in vacuo followed by flash chromatography (1:2 hexanes/ethyl acetate) afforded 12 as a white solid (74 mg, 73%): mp 75-76 °C; ¹H NMR (300 MHz, CDCl₃) δ 3.96 (t, J =7.8 Hz, 1H), 3.76 (dd, J = 11.7, 3.7 Hz, 1H), 3.67 (dd, J = 11.7, 4.4 Hz, 1H), 3.64 (d, J = 11.5 Hz, 1H), 3.58 (t, J = 3.9 Hz, 1H), 3.35 (d, J = 11.5 Hz, 1H), 3.11 (bs, 1H), 2.66 (bs, 2H), 2.05 (m, 1H), 1.90 (m, 2H), 1.59 (m, 1H), 1.20 (s, 3H), 1.12 (s, 3H), 0.91 (s, 9H), 0.11 (s, 3H), 0.10 (s, 3H); $^{13}\mathrm{C}$ NMR (75 MHz, CDCl₃) δ 85.9, 83.5, 77.1, 73.0, 67.2, 64.1, 34.0, 26.5, 25.8, 24.1, 21.6, 18.1, -4.5, -4.7; IR (neat) 3396 cm⁻¹; HRMS calcd for C₁₆H₃₄SiO₅ + H 335.2254, found 335.2250; LRMS (EI) m/z 335. Anal. Calcd for C₁₆H₃₄SiO₅: C, 57.44; H, 10.24; Found: C, 57.31; H, 10.20.

 $(1''R^*, 1'R^*, 2S^*, 5R^*)$ -5-(1', 2'-Dihydroxy-1'-methylethyl)-5methyl-2-[2"-[(tert-butyldimethylsilyl)oxy]-1"-hydroxyethyl]tetrahydrofuran (13). A solution of the triol 12 (450 mg, 1.35 mmol) in CH₂Cl₂ (40 mL) was treated with DBU (205 mg, 1.35 mmol) and stirred at rt overnight under argon. Saturated NH4Cl solution was added, and the mixture was extracted with CH₂Cl₂. The organic phase was washed with brine and dried over MgSO₄. Concentration in vacuo followed by flash chromatography (3:1 hexanes/acetone) afforded recovered starting material (70 mg, 16%) and 13 as a colorless oil (370 mg, 82%): ¹H NMR (300 MHz, CDCl₃) δ 3.98 (t, J = 7.3 Hz, 1H), 3.73 (dd, J = 9, 2.9 Hz, 1H), 3.67 (d, J = 11.5 Hz, 1H), 3.60 (t, J = 9 Hz, 1H), $3.54 \, (dd, J = 9, 2.9 \, Hz, 1H)$, $3.31 \, (d, J = 11.5 \, Hz, 1H)$, $3.3-3.0~(bs,~3H),~2.23~(m,~1H),~1.90~(m,~2H),~1.62~(m,~1H),~1.11~(s,~3H),~1.09~(s,~3H),~0.87~(s,~9H),~0.06~(s,~6H);~^{13}C$ NMR (75 MHz, CDCl₃) & 85.1, 83.0, 76.2, 73.2, 67.1, 63.5, 35.3, 26.8, 25.8, 22.7, 21.9, 18.2, -5.5; IR (neat) 3416 cm⁻¹; HRMS calcd for C₁₆H₃₄-SiO₅ + H 335.2254, found 335.2261; LRMS (CI) m/z 335.

(1"R*,1'R*,2S*,5R*)-5-[(1-Methylethylidene)dioxy]-1'-methylethyl]-5-methyl-2-[2"-[(tert-butyldimethylsilyl)oxy]-1"hydroxyethyl]tetrahydrofuran. A solution of 13 (197 mg, 0.59 mmol), $Me_2C(OMe)_2$ (1.3 mL), and PPTS (20 mg) in CH₂-Cl₂ (10 mL) was stirred at rt under argon for 4 h. Saturated NaHCO₃ solution was added, the mixture was extracted with CH₂Cl₂, and the organic phase was dried over MgSO₄. Concentration in vacuo followed by flash chromatography (4:1 hexanes/ ethyl acetate) afforded the acetonide as a colorless oil (212 mg, 96%): ¹H NMR (300 MHz, CDCl₃) δ 3.97 (d, J = 8.5 Hz, 1H), 3.93 (t, J = 7.8 Hz, 1H), 3.69 (d, J = 8.5 Hz, 1H), 3.68 (dd, J = 7.8 Hz, 1H), 3.8 Hz, 1H), 3.68 (dd, J = 7.8 Hz, 1H), 3.8 Hz,6.8, 5.1 Hz, 1H), 3.59 (dd, J = 10, 6.8 Hz, 1H), 3.49 (dd, J = 6.8, 5.1 Hz, 1H), 2.85 (bs, 1H), 2.08 (m, 1H), 1.94 (m, 1H), 1.80 (m, 1H), 1.60 (dt, J = 12.2, 8.1 Hz, 1H), 1.39 (s, 3H), 1.37 (s, 3H), 1.26 (s, 3H), 1.16 (s, 3H), 0.87 (s, 9H), 0.05 (s, 6H); ¹³C NMR (75) MHz, CDCl₃) & 109.6, 84.5, 82.3, 82.1, 72.3, 64.1, 34.8, 27.0, 26.7, 25.8, 22.1, 21.5, 18.2, -5.4, -5.5; IR (neat) 3490 cm⁻¹; HRMS calcd for $\rm C_{19}H_{38}SiO_5+H$ 375.2567, found 375.2543; LRMS (CI) m/z 375. Anal. Calcd for C₁₉H₃₈O₅Si: C, 60.92; H, 10.22. Found: C, 60.74; H, 10.35.

(1"R*,1'R*,2S*,5R*)-5-[1',2'-[(1-Methylethylidene)dioxy]-1'-methylethyl]-5-methyl-2-[2"-[(tert-butyldimethylsilyl)oxy] - 1'' - (methane sulfony loxy) ethyl] tetrahydrofur an (14).A solution of the acetonide (210 mg, 0.56 mmol), MsCl (332 mg, 2.8 mmol), and DMAP (1 mg) in pyridine (4 mL) was stirred overnight under argon. The solution was poured into water and extracted with ether. The organic phase was washed with 10% HCl, saturated NaHCO₃, and brine and dried over MgSO₄. Concentration in vacuo followed by flash chromatography (9:1 hexanes/ethyl acetate) afforded 14 as a colorless oil (230 mg, 91%): ¹H NMR (300 MHz, CDCl₃) δ 4.54 (dd, J = 7.3, 3.2 Hz, 1H), 3.97 (d, J = 8.5 Hz, 1H), 3.76-3.94 (m, 3H), 3.67 (d, J = 3.5 Hz)8.5 Hz, 1H), 3.09 (s, 3H), 2.75–2.10 (m, 3H), 1.65 (m, 1H), 1.35 (s, 6H), 1.22 (s, 3H), 1.21 (s, 3H), 0.87 (s, 9H), 0.06 (s, 3H), 0.05 (s, 3H); 13 C NMR (75 MHz, CDCl₃) δ 109.5, 89.5, 83.3, 82.0, 81.9, 72.3, 62.9, 38.9, 35.1, 26.9, 26.9, 26.8, 25.8, 22.8, 20.9, 18.3, -5.5, -5.6; HRMS calcd for C₂₀H₄₀O₇SSi - CH₃ 437.2029, found 437.2012; LRMS (CI) m/z 470, 453. Anal. Calcd for C20H40O7-SSi: C, 53.06; H, 8.91. Found: C, 52.89; H, 9.20.

 $(1''R^*, 1'R^*, 2S^*, 5R^*)$ -5-[1', 2'-[(1-Methylethylidene)dioxy]-1'-methylethyl]-5-methyl-2-(1'', 2''-epoxyethyl)tetrahydrofuran (15). A stirred solution of 14 (183 mg, 0.40 mmol) in THF (4 mL) was treated with Bu₄NF (0.45 mL of a 1.0 M solution in THF, 0.45 mmol). After 45 min solid K₂CO₃ (700 mg) was added and the mixture was stirred for a further 7 h. Water was added, and the aqueous phase was extracted with ethyl acetate. The organic phase was washed with brine and dried over Na₂SO₄. Concentration *in vacuo* followed by flash chromatography (4:1 hexanes/ethyl acetate) afforded **15** as a colorless oil (96 mg, 98%): ¹H NMR (300 MHz, CDCl₃) δ 4.02 (d, J = 8.5 Hz, 1H), 3.95 (t, J = 6.6 Hz, 1H), 3.70 (d, J = 8.5 Hz, 1H), 3.00 (dd, J = 4.1, 2.7 Hz, 1H), 2.70 (t, J = 4.6 Hz, 1H), 2.56 (dd, J = 4.9, 2.7 Hz, 1H), 1.30 (m, 1H), 1.70–1.86 (m, 2H), 1.52 (m, 1H), 1.36 (s, 6H), 1.23 (s, 3H), 1.18 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 109.4, 82.2, 82.1, 81.9, 72.9, 56.6, 44.3, 32.8, 26.9, 26.8, 23.1, 20.0; HRMS calcd for C₁₃H₂₂O₄ – CH₃ 227.1283; found 227.1284; LRMS (CI) *m*/z 260. Anal. Calcd for C₁₃H₂₂O₄: C, 64.43, H, 9.15. Found: C, 64.08; H, 9.03.

(1"R*,1'R*,2S*,5R*)-5-[1'-Hydroxy-2'-(methanesulfonyloxy)-1'-methylethyl]-5-methyl-2-[2"-[(tert-butyldimethylsilyl)oxy]-1".(methanesulfonyloxy)ethyl]tet. rahydrofuran (16). A solution of 13 (240 mg, 0.72 mmol), MsCl (183 mg, 1.58 mmol), and DMAP (1 mg) in pyridine (5 mL) was stirred overnight under argon. The solution was poured into water and extracted with ether. The organic phase was washed with 10% HCl, saturated NaHCO3, and brine and dried over MgSO₄. Concentration in vacuo followed by flash chromatography (7:3 hexanes/ethyl acetate) afforded 16 as a colorless oil (300 mg, 85%): ¹H NMR (300 MHz, CDCl₃) δ 4.53 (t, J = 5.6Hz, 1H), 4.11 (d, J = 10.5 Hz, 1H), 4.03 (d, J = 10.5 Hz, 1H), 3.93 (t, J = 7.1 Hz, 1H), 3.72 (d, J = 5.6 Hz, 2H), 3.10 (bs, 1H), 3.08 (s, 3H), 3.04 (s, 3H), 1.97 (m, 3H), 1.73 (m, 1H), 1.27 (s, 3H), 1.18 (s, 3H), 0.86 (s, 9H), 0.06 (s, 3H), 0.05 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) & 89.3, 82.3, 81.9, 73.8, 71.8, 62.8, 39.0, 37.2, 35.7, 25.9, 25.8, 21.8, 21.5, 18.2, -5.5, -5.6; IR (neat) 3500 cm⁻¹; HRMS calcd for $C_{18}H_{38}S_2SiO_9 + H$ 491.1805, found 491.1804; LRMS (CI) m/z 491.

(1"R*,1'R*,2S*,5R*)-5-(1',2'-Epoxy-1'-methylethyl)-5-methvl-2-[2"-[(tert-butyldimethylsilyl)oxy]-1"-(methanesulfonyloxy)ethyl]tetrahydrofuran (17). A mixture of 16 (250 mg, 0.51 mmol) and K₂CO₃ (300 mg, 2.17 mmol) in methanol (10 mL) was stirred at rt for 1.5 h. Water was added, and the solution was extracted with ether. The combined organic phase was washed with brine and dried over Na₂SO₄. Concentration in vacuo followed by flash chromatography (85:15 hexanes/ethyl acetate) afforded 16 as a colorless oil (195 mg, 97%): ¹H NMR (300 MHz, CDCl₃) δ 4.54 (dd, J = 7.6, 3.7 Hz, 1H), 3.92-3.74 (m, 3H), 3.12 (s, 3H), 2.75 (d, J = 4.9 Hz, 1H), 2.56 (d, J = 4.9Hz, 1H), 2.14-1.62 (m, 4H), 1.31 (s, 3H), 1.22 (s, 3H), 0.87 (s, 9H), 0.06 (s, 3H), 0.06 (s, 3H); $^{13}\mathrm{C}$ NMR (75 MHz, CDCl₃) δ 89.2, 83.5, 80.0, 62.8, 57.2, 51.9, 38.9, 35.2, 27.7, 25.8, 22.8, 18.3, 17.7, -5.5, -5.6; HRMS calcd for C17H34O6SSi 395.1923, found 395.1928; LRMS (CI) m/z 395. Anal. Calcd for C17H34O6SSi: C, 51.74, H, 8.68. Found: C, 52.07; H, 8.81.

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Supplementary Material Available: ¹H and ¹³C NMR spectra of compounds for which no elemental analysis was obtained (4 pages). This material is contained in libraries on microfiche, immediately follows this article in the microfilm version of the journal, and can be ordered from the ACS; see any current masthead page for ordering information.